IN THE UNITED STATES COURT OF APPEALS FOR THE DISTRICT OF COLUMBIA CIRCUIT

.

NATIONAL WASTE & RECYCLING ASSOCIATION	• : :	
Petitioner,	:	
ν.	:	Case No. 24-1321
UNITED STATES ENVIRONMENTAL	:	
PROTECTION AGENCY, and MICHAEL S. REGAN, in his official	:	
Capacity as Administrator, United States Environmental Protection Agency,	:	
Respondents.	:	

PETITION FOR REVIEW

Pursuant to Section 307(b) of the Clean Air Act, 42 U.S.C. § 7607(b), Section 702 of the Administrative Procedure Act, 5 U.S.C. § 702, Rule 15(a) of the Federal Rules of Appellate Procedure, and Rule 15(a) of the D.C. Circuit Rules, Petitioner National Waste & Recycling Association hereby petitions this Court for review of the final action taken by Respondents United States Environmental Protection Agency (EPA) and Administrator Michael Regan, titled *"Final Emissions Factors for AP-42 Chapter 2, Section 4 - Municipal Solid Waste Landfills,"* dated August 15, 2024, *available at* https://www.epa.gov/air-emissions-factors-andquantification/final-emissions-factors-ap-42-chapter-2-section-4. Dated: October 14, 2024

Respectfully submitted,

/s/ Carol F. McCabe

Carol F. McCabe Kelly A. Hanna MANKO, GOLD, KATCHER & FOX Three Bala Plaza East Suite 700 Bala Cynwyd, PA 19004 Phone: 484-430-2304 Fax: 484-430-5711 cmccabe@mankogold.com khanna@mankogold.com

Matthew W. Morrison Mark C. Talty PILLSBURY WINTHROP SHAW PITTMAN 200 Seventeenth Street, NW Washington, DC 20036 Phone: 202-663-8000 Fax: 202-663-8007 matthew.morrison@pillsburylaw.com mark.talty@pillsburylaw.com

Counsel for Petitioner National Waste & Recycling Association

IN THE UNITED STATES COURT OF APPEALS FOR THE DISTRICT OF COLUMBIA CIRCUIT

	•		
NATIONAL WASTE & RECYCLING ASSOCIATION	:		
	:		
Petitioner,	:		
	:		
ν.	:	Case No. 24	
	:		
UNITED STATES ENVIRONMENTAL	:		
PROTECTION AGENCY, and	:		
MICHAEL S. REGAN, in his official	:		
Capacity as Administrator, United States	:		
Environmental Protection Agency,	:		
	:		
Respondents.	:		
	:		

RULE 26.1 DISCLOSURE STATEMENT

Pursuant to Rule 26.1 of the Federal Rules of Appellate Procedure and Circuit Rule 26.1, Petitioner National Waste & Recycling Association hereby files the following corporate disclosure statement:

The National Waste & Recycling Association (the Association) is a trade association that represents private sector waste and recycling companies in the United States, and manufacturers and service providers who do business with those companies. Its members operate in all 50 states and the District of Columbia. The Association provides leadership, education, research, advocacy, and safety expertise to promote North American waste and recycling industries, serve as their voice, and create a climate where members prosper and provide safe, economically sustainable, and environmentally sound services. The Association has no parent corporation, and no publicly held company has 10% or greater ownership in the Association. Dated: October 14, 2024

Respectfully submitted,

/s/ Carol F. McCabe

Carol F. McCabe Kelly A. Hanna MANKO, GOLD, KATCHER & FOX Three Bala Plaza East Suite 700 Bala Cynwyd, PA 19004 Phone: 484-430-2304 Fax: 484-430-5711 cmccabe@mankogold.com khanna@mankogold.com

Matthew W. Morrison Mark C. Talty PILLSBURY WINTHROP SHAW PITTMAN 200 Seventeenth Street, NW Washington, DC 20036 Phone: 202-663-8000 Fax: 202-663-8007 matthew.morrison@pillsburylaw.com mark.talty@pillsburylaw.com

Counsel for Petitioner National Waste & Recycling Association

CERTIFICATE OF SERVICE

Pursuant to Rule 15(c) and 25 of the Federal Rules of Appellate Procedure and Circuit

Rule 15(a) and 25, I hereby certify that on October 14, 2024, I will cause copies of the foregoing

Petition for Review and Rule 26.1 Disclosure Statement to be served via UPS overnight delivery

on each of the following:

The Honorable Michael S. Regan Administrator U.S. Environmental Protection Agency Office of the Administrator (Mail Code 1101A) 1200 Pennsylvania Avenue NW Washington, D.C. 20460

Associate General Counsel for the Air and Radiation Law Office U.S. Environmental Protection Agency Office of General Counsel (Mail Code 2344A) 1200 Pennsylvania Avenue NW Washington, DC 20464

The Honorable Merrick B. Garland Attorney General of the United States U.S. Department of Justice 950 Pennsylvania Avenue NW Washington, D.C. 20530-0001

Correspondence Control Unit U.S. Environmental Protection Agency Office of General Counsel (Mail Code 2310A) 1200 Pennsylvania Avenue NW Washington, D.C. 20460

Dated: October 14, 2024

Respectfully submitted,

<u>/s/ Carol F. McCabe</u> Carol F. McCabe

Counsel for Petitioner National Waste & Recycling Association

Attachment 1

2.4 Municipal Solid Waste Landfills

Disclaimer: Emission factors in AP-42 are neither EPA-recommended emission limits (e.g., best available control technology or BACT, or lowest achievable emission rate or LAER) nor standards (e.g., National Emission Standard for Hazardous Air Pollutants or NESHAP, or New Source Performance Standards or NSPS). Use of these factors as source-specific permit limits and/or as emission regulation compliance determinations is **NOT** recommended by EPA. Because emission factors essentially represent an average of a range of emission rates, approximately half of the subject sources are expected to have emission rates greater than the emission factor, and the other half are expected to have emission factor. As such, EPA does not recommend using emission factors as limits or standards. This could cause, for example, a permit limit using an AP-42 emission factor resulting in approximately half of the sources being in noncompliance. We recommend source testing be done for the best possible emission values. For more information on the use of emission factors, please refer to the <u>AP-42 Introduction</u>.

2.4.1 General¹⁻⁴

A municipal solid waste (MSW) landfill unit is a discrete area of land or an excavation that receives household waste, and that is not a land application unit, surface impoundment, injection well, or waste pile. An MSW landfill unit may also receive other types of wastes, such as commercial solid waste, nonhazardous sludge, and industrial solid waste. The municipal solid waste types potentially accepted by MSW landfills include (most landfills accept only a few of the following categories):

- MSW,
- Household hazardous waste,
- Municipal sludge,
- Municipal waste combustion ash,
- Infectious waste,
- Waste tires,
- Industrial non-hazardous waste,
- Conditionally exempt small quantity generator (CESQG) hazardous waste,
- Construction and demolition waste,
- Agricultural wastes,
- Oil and gas wastes, and
- Mining wastes.

In the United States in 2018, approximately 50% of solid waste was landfilled, 12% was combusted for energy recovery, and 32% was recycled or composted.⁵ There were an estimated 1,274 active MSW landfills in the United States in 2021. These landfills were estimated to receive 339 million megagrams (Mg) (373 million tons) of waste annually.⁶ In 1998, 55 to 65% of MSW was reported as household waste and 35 to 45% of MSW was reported as commercial waste.⁷

References for this AP-42 section are available electronically <u>here.</u> The reader is referred to Sections <u>13.2.2</u> (<u>Unpaved Roads</u>), and <u>11.2.4 (Heavy Construction Operations</u>) of the Electronic AP-42: Compilation of Air Emissions Factors from Stationary Sources, and <u>Section II-7 (Construction Equipment) of Volume II</u>, of the AP-42 document for determination of associated fugitive dust and exhaust emissions from these emission sources at MSW landfills. In addition to this, <u>Section 3.1 (Stationary Gas Turbines</u>) and <u>Section 13.5 (Industrial Flares</u>) of the Electronic AP-42: Compilation of Air Emissions Factors from Stationary Sources also contains emission factors for landfill gas (LFG) fired turbines and open flares, respectively.

2.4.2 Process Description^{2,8}

There are three major designs for municipal landfills. These are the area, trench, and ramp methods. All of these methods utilize a three-step process, which includes spreading the waste, compacting the waste, and covering the waste with soil. The trench and ramp methods are not commonly used and are not the preferred methods when liners and leachate collection systems are utilized or required by law. The area fill method

involves placing waste on the ground surface or landfill liner, spreading it in layers, and compacting with heavy equipment. A daily soil cover is spread over the compacted waste. The trench method entails excavating trenches designed to receive a day's worth of waste. The soil from the excavation is often used for cover material and wind breaks. The ramp method is typically employed on sloping land, where waste is spread and compacted similar to the area method, however, the cover material obtained is generally from the front of the working face of the filling operation.

Modern landfill design often incorporates liners constructed of soil (i.e., recompacted clay), or synthetics (i.e., high density polyethylene), or both to provide an impermeable barrier to leachate (i.e., water that has passed through the landfill) and gas migration from the landfill.

2.4.3 Control Technology 1,2,9

The Resource Conservation and Recovery Act (RCRA) Subtitle D regulations promulgated on October 9, 1991, require that the concentration of methane generated by MSW landfills not exceed 25% of the lower explosive limit (LEL) in on-site structures, such as scale houses, or the LEL at the facility property boundary.

The original New Source Performance Standards (NSPS) and Emission Guidelines (EG) for air emissions from MSW landfills for certain new and existing landfills were published in the Federal Register on March 12, 1996. Since then, the MSW NSPS/EG were updated on August 29, 2016. Additionally, a National Emission Standard for Hazardous Air Pollutants (NESHAP) was promulgated on January 16, 2003, and the residual risk and technology review (RTR) was promulgated on March 26, 2020, with technical corrections to the RTR promulgated on February 3, 2022. A history of MSW landfills can be found on the EPA's <u>Municipal Solid Waste Landfills: New Source</u> Performance Standards (NSPS), Emission Guidelines (EG) and Compliance Times website. The NSPS, EG, and NESHAP for MSW landfills are similar in that they regulate emissions of landfill gas using non-methane organic compounds (NMOC) as an estimate for VOC emissions. These regulations established a design capacity of 2.5 million Mg (2.75 million tons) and 2.5 million cubic meters and NMOC emission rate thresholds that if exceeded require landfills to install a gas collection and control system (GCCS). Control systems require: (1) a well-designed and well-operated GCCS, and (2) a control device capable of reducing NMOCs in the collected gas by 98 weight-percent.

Landfill gas (LFG) collection systems, also referred to as GCCS, are either active or passive systems. Active collection systems provide a pressure gradient in order to extract LFG by use of mechanical blowers or compressors. Passive systems allow the natural pressure gradient created by the increase in pressure created by LFG generation within the landfill to mobilize the gas for collection.

LFG control and treatment options include (1) combustion of the LFG, and (2) purification of the LFG. Combustion techniques include techniques that do not recover energy (i.e., flares and thermal incinerators), and techniques that recover energy (i.e., gas turbines and internal combustion engines) and generate electricity from the combustion of the LFG. Boilers can also be employed to recover energy from LFG in the form of steam. Flares involve an open combustion process that requires oxygen for combustion and can be open or enclosed. Thermal incinerators heat an organic chemical to a high enough temperature in the presence of sufficient oxygen to oxidize the chemical to carbon dioxide (CO₂) and water. Purification techniques can also be used to process raw landfill gas to pipeline quality natural gas by using adsorption, absorption, and membranes.

2.4.4 Emissions^{2,10}

Methane (CH₄) and CO₂ are the primary constituents of landfill gas and are produced by microorganisms within the landfill under anaerobic conditions. Transformations of CH₄ and CO₂ are mediated by microbial populations that are adapted to the cycling of materials in anaerobic environments. Landfill gas generation, including rate and composition, proceeds through four phases. The first phase is aerobic [i.e., with oxygen (O₂) available] and the primary gas produced is CO₂. The second phase is characterized by O₂ depletion, resulting in an anaerobic environment, where large amounts of CO₂ and some hydrogen (H₂) are produced. In the third phase, CH₄ production begins, with an accompanying reduction in the amount of CO₂ produced. Nitrogen (N₂) content is initially high in landfill gas in the first phase and declines sharply as the landfill proceeds through the second and

third phases. In the fourth phase, gas production of CH₄, CO₂, and N₂ becomes fairly steady. The total time and phase duration of gas generation varies with landfill conditions (i.e., waste composition, design management, and anaerobic state).

Typically, LFG also contains a small amount of NMOC. This NMOC fraction often contains various organic hazardous air pollutants (HAP), greenhouse gases (GHG), and compounds associated with stratospheric ozone depletion. The NMOC fraction also contains volatile organic compounds (VOC). The weight fraction of VOC can be determined by subtracting the weight fractions of individual compounds that are non-photochemically reactive (i.e., negligibly reactive organic compounds as defined in 40 CFR 51.100).

Other emissions associated with MSW landfills include combustion products from LFG control and utilization equipment (i.e., flares, engines, turbines, and boilers). These include carbon monoxide (CO), oxides of nitrogen (NO_x), sulfur dioxide (SO₂), hydrogen chloride (HCl), particulate matter (PM) and other combustion products (including HAPs). PM emissions can also be generated in the form of fugitive dust created by mobile sources (i.e., garbage trucks) traveling along paved and unpaved surfaces. The reader should consult AP-42 Volume I Sections 13.2.1 and 13.2.2 for information on estimating fugitive dust emissions from paved and unpaved roads.

The rate of emissions from a landfill is governed by gas production and transport mechanisms. Production mechanisms involve the production of the emission constituent in its vapor phase through vaporization, biological decomposition, or chemical reaction. Transport mechanisms involve the transportation of a volatile constituent in its vapor phase to the surface of the landfill, through the air boundary layer above the landfill, and into the atmosphere. The three major transport mechanisms that enable transport of a volatile constituent in its vapor phase are diffusion, convection, and displacement.

2.4.4.1 Uncontrolled Emissions

EPA notes that fugitive emissions include those emissions not gathered by collection devices (uncollected). Here, fugitive, uncollected, and uncontrolled emissions are synonymous terms. One way to estimate uncontrolled emissions of a pollutant in landfill gas is to begin by determining the annual volume of landfill methane generation, accounting for air infiltration as necessary, and to use the ideal gas law to provide pollutant mass per year. Methane generation may be estimated for individual landfills by multiplying the result of Equation HH-1, found at 40 CFR 98.343(a)(1), by 1474.83 to obtain methane generation for the reporting year for which emissions are calculated in terms of cubic meters per year. The equation is as follows:

$$Q_{CH4} = G_{CH4} \frac{1000}{(0.0192)(35.3147)} = G_{CH4} \times 1474.83 \tag{1}$$

where:

=	Methane generation rate for the reporting year, m ³ /yr;
=	Result of Equation HH-1, metric tons CH ₄ /yr;
=	Conversion, kilograms to metric tons;
=	Density of methane at 60° F and 14.7 psia, kg/ft ³ ; and
=	conversion, ft ³ to m ³ .
	= =

The Landfill Gas Emissions Model (LandGEM) is an automated estimation tool with a Microsoft Excel interface that can be used to estimate generation or emissions rates for total landfill gas, methane, carbon dioxide, and NMOCs, and individual air pollutants from municipal solid waste landfills. Version 3.1, available from the following EPA website: https://www.epa.gov/system/files/other-files/2023-12/landgem-v3.1beta-dec-2023.xlsm, was updated in December 2023 and includes Equation HH-1 from 40 CFR 98.343(a)(1) and its selectable parameters, as well as the theoretical first-order kinetic model of methane production found in LandGEM Version 3.03. Note that the regulatory defaults for the equations in LandGEM Version 3.03 must be used when required by programs such as 40 CFR Part 60 subpart XXX, 40 CFR Part 60 Subpart Cf (Emission Guideline), or 40 CFR Part 63 AAAA.

When gas generation reaches steady state conditions, LFG consists of approximately 40% by volume CO_2 , 55% CH_4 , 5% N_2 (and other gases), and trace amounts of NMOCs. Therefore, the estimate derived for CH_4 generation using LandGEM can also be used to represent CO_2 generation. Addition of the CH_4 and CO_2 emissions will yield an estimate of total landfill gas emissions. If site-specific information is available to suggest that the CH_4 content of landfill gas is not 55%, then the site-specific information should be used, and the CO_2 emission estimate should be adjusted accordingly.

For landfills, volatile organic compound (VOC) emissions are equivalent to NMOC emissions minus the emissions from compounds with low to no photochemical reactivity. Predominant compounds with low to no photochemical reactivity found in landfills include methyl chloroform, acetone, methylene chloride, tetrachloroethylene, chlorodifluoromethane, dichlorodifluoromethane, and ethane. When the contribution of emissions from these compounds with low to no photochemical reactivity is low, then NMOC emissions are a good surrogate for VOC emissions. Recent data review shows that the contribution of these seven predominant compounds with low to no photochemical reactivity to be less than 6.5% of NMOC.¹¹⁻³⁸

Most of the NMOC emissions result from the volatilization of organic compounds contained in the landfilled waste. Small amounts may be created by biological processes and chemical reactions within the landfill. The current version of LandGEM contains a regulatory default value for total NMOC of 4,000 ppmv, expressed as hexane. The regulatory default value for NMOC concentration was developed for regulatory compliance purposes (40 CFR Part 60, Subpart XXX) and to provide the most cost-effective default values on a national basis. For emissions inventory purposes, site-specific information should be taken into account when determining the total NMOC concentration. In the absence of site-specific information, a value of 2,400 ppmv as hexane is suggested for landfills known to have co-disposal of MSW and non-residential waste. If the landfill is known to contain only MSW or have very little organic commercial/industrial wastes, then for default values before 1992, 600 ppmv as hexane should be used, and for default values on and after 1992, 550 ppmv as hexane should be used. In addition, as with the landfill model defaults, the regulatory default value for NMOC content must be used in order to comply with the NSPS/Emission Guideline. According to NSPS (40 CFR Part 60, Subpart XXX) and Emission Guideline (40 CFR Part 60, Subpart Cf), the landfills with annual NMOC emissions greater than 34 megagrams must consider further emission measurement efforts or installation of a gas collection system.

Before a default pollutant concentration is used (e.g. from Table 2.4-1), it must be reviewed for potential air infiltration correction. from Table 2.4-1), it must be reviewed for potential air infiltration correction. Air infiltration can occur by two different mechanisms: LFG sample dilution, and air intrusion into the landfill. These corrections require site-specific data for the LFG CH₄, CO₂, nitrogen (N₂), and oxygen (O₂) content. If the ratio of N₂ to O₂ is less than or equal to 4.0 (as found in ambient air), then the total pollutant concentration is adjusted for sample dilution by assuming that CO₂ and CH₄ are the primary (100%) constituents of landfill gas, and the following equation is used:

$$C_P(ppmv) (corrected for air infiltration) = \frac{C_P(ppmv)(1 \times 10^6)}{C_{CO_2}(ppmv) + C_{CH_4}(ppmv)}$$
(2)

where:

C_P	=	Concentration of pollutant P in landfill gas (e.g., NMOC as hexane), ppmv;
C_{CO_2}	=	CO_2 concentration in landfill gas, ppmv;
C_{CH_4}	=	CH_4 Concentration in landfill gas, ppmv; and
1x10 ⁶	=	Constant used to correct concentration of P to units of ppmv.

If the ratio of N₂ to O₂ concentrations (i.e., C_{N_2} , C_{O_2}) is greater than 4.0, then the total pollutant concentration should be adjusted for air intrusion into the landfill by using equation 2 and adding the concentration of N₂ (i.e., C_{N_2}) to the denominator. Values for C_{CO_2} , C_{CH_4} , C_{N_2} , and C_{O_2} , can usually be found in the source test report for the landfill along with the total pollutant concentration data.

To estimate emissions of NMOC or other landfill gas constituents, the following equation should be used:

$$Q_P = \frac{1}{F} Q_{CH_4} \times \frac{C_P}{(1 \times 10^6)} \tag{3}$$

where:

Q_P	=	Emission rate of pollutant P (e.g., NMOC), m³/yr;
F	=	Fraction by volume of CH4 in landfill gas from measurement data for the current
		reporting year, if available (fraction, dry basis, corrected to 0% oxygen); otherwise,
		use the default of 0.5;
Q_{CH_4}	=	CH₄ generation rate, m³/yr (from equation 1 or LandGEM); and
Cp	=	Concentration of P in landfill gas, ppmv.

Uncontrolled mass emissions per year of total NMOC (as hexane), CO₂, CH₄, and speciated organic and inorganic compounds can be estimated by the following equation:

$$UM_P = Q_P \times \frac{MW_P \times 1 \ atm}{(8.205 \times 10^{-5} \ \frac{m^3 \ atm}{g \ mol^{\circ} \kappa})(1000 \ \frac{g}{kg})(273 + T^{\circ} \kappa)}$$
(4)

where:

UM_P	=	Uncontrolled mass emissions of pollutant P (e.g., NMOC), kg/yr;
MW_P	=	Molecular weight of P, g/gmol (e.g., 86.18 for NMOC as hexane);
Q_P	=	NMOC emission rate of P, m ³ /yr; and
Т	=	Temperature of landfill gas, °C.

This equation assumes that the operating pressure of the system is approximately 1 atmosphere. If the temperature of the landfill gas is not known, a temperature of 25°C (77°F) is recommended.

Uncontrolled default concentrations of speciated organics along with some inorganic compounds are presented in Table 2.4-1. These default concentrations have already been corrected for air infiltration and can be used as input parameters to equation 3 for estimating speciated emissions from landfills when site-specific data are not available. An analysis of the data, based on the co-disposal history (with non-residential wastes) of the individual landfills from which the concentration data were derived, indicates that for benzene, NMOC, and toluene, there is a difference in the uncontrolled concentrations. Table 2.4-2 presents the corrected concentrations for benzene, NMOC, and toluene to use based on the site's co-disposal history.

It is important to note that the compounds listed in Tables 2.4-1 and 2.4-2 are not the only compounds likely to be present in LFG. The listed compounds are those that were identified through a review of the available literature. The reader should be aware that additional compounds are likely present, such as those associated with consumer or industrial products. Given this information, extreme caution should be exercised in the use of the default VOC weight fractions and concentrations given at the bottom of Table 2.4-2. These default VOC values are heavily influenced by the ethane content of the LFG. Available data have shown that there is a range of over 1,500 ppmv in LFG ethane content among landfills.

2.4.4.2 Controlled Emissions

Emissions from landfills are typically controlled by installing a gas collection system and combusting the collected gas through the use of internal combustion engines, flares, or turbines. Gas collection systems are not 100% efficient in collecting landfill gas, so emissions of CH₄ and NMOC at a landfill with a gas recovery system still occur. To estimate controlled emissions of CH₄, NMOC, and other constituents in landfill gas, the collection efficiency of the system should first be estimated. Different models exist to provide a better estimate for uncontrolled emissions that are more appropriate for emission inventory development (e.g. LandGEM) that incorporates waste degradation parameters which include the potential to generate methane and the waste degradation decay rate. The potential to generate methane is a function of waste type, and age of waste. The waste degradation decay rate is a function of waste type, age of waste, and waste moisture. Waste moisture might be changed by leachate recirculation and rainfall rates. Higher collection efficiencies may be achieved at some sites (i.e., those engineered to control gas emissions). If site-specific collection efficiencies are available (i.e., through a comprehensive surface sampling program), then they should be used instead. If a user lacks sitespecific collection efficiencies, use the appropriate values in Table HH-3 to Subpart HH of Part 98 - Landfill Gas Collection Efficiencies for calculations. See section III.T.2 of the preamble of the GHGRP final rule (89 FR 31853 April 25, 2024) for more information on the finalized collection efficiencies in Table HH-3 and default collection efficiency.

Controlled emission estimates should also take into consideration the control efficiency of the control device. Control efficiencies based on test data for the combustion of CH₄, NMOC, and some speciated organics with differing control devices are presented in Table 2.4-3. Emissions from the control devices need to be added to the uncollected emissions to estimate total controlled emissions.

Controlled CH₄, NMOC, and speciated emissions can be calculated with equation 5. It is assumed that the landfill GCCS operates 100% of the time. Minor durations of system downtime associated with routine maintenance and repair (i.e., 5 to 7%) should not appreciably affect emission estimates. The first term in equation 5 accounts for emissions from uncollected landfill gas, while the second term accounts for emissions of the pollutant that were collected but not combusted in the control or utilization device:

$$CM_P = \left[UM_P \times \left(1 - \frac{\eta_{col}}{100} \right) \right] + \left[UM_P \times \frac{\eta_{col}}{100} \times \left(1 - \frac{\eta_{cnt}}{100} \right) \right]$$
(5)

where:

CM_P	=	Controlled mass emissions of pollutant P, kg/yr;
UM_P	=	Uncontrolled mass emissions of P, kg/yr (from equation 4 or LandGEM);
η_{col}	=	Collection efficiency of the landfill gas collection system, percent; and
η_{cnt}	=	Control efficiency of the landfill gas control or utilization device, percent.

Emission factors for the secondary compounds, CO and NO_x, exiting the control device are presented in Tables 2.4-4 and 2.4-5. For convenience, emission factors are also presented for NMOC, although most of this NMOC is presumed to be from incomplete combustion of NMOC generated from the landfill rather than NMOC generated by combustion. These default values can be used when equipment vendor guarantees are not available.

Consistent with the language in the Introduction to AP-42, using source-specific data is preferred for estimating a source's emissions, while controlled emissions of CO_2 and SO_2 are best estimated using site-specific landfill gas constituent concentrations, along with mass balance methods.³⁹ If site-specific data are not available, the data in Tables 2.4-1 through 2.4-3 can be used with the mass balance methods that follow.

Controlled CO₂ emissions include emissions from the CO₂ component of landfill gas (equivalent to

uncontrolled emissions) and additional CO_2 formed during the combustion of landfill gas. The bulk of the CO_2 formed during landfill gas combustion comes from the combustion of the CH_4 fraction. Small quantities will be formed during the combustion of the NMOC fraction; however, this typically amounts to less than 1% of total CO_2 emissions by weight. Also, the formation of CO through incomplete combustion of landfill gas will result in small quantities of CO_2 not being formed. This contribution to the overall mass balance is also very small and does not have a significant impact on overall CO_2 emissions.³⁹

The following equation, which assumes a 100% combustion efficiency for CH_4 , can be used to estimate CO_2 emissions from controlled landfills:

$$CM_{CO_2} = UM_{CO_2} + (UM_{CH_4} \times \frac{\eta_{col}}{100} \times 2.75)$$
 (6)

where:

CM_{CO_2} =	Controlled mass emissions of CO ₂ , kg/yr;
$UM_{CO_2} =$	Uncontrolled mass emissions of CO_2 , kg/yr (from equation 4 or LandGEM);
$UM_{CH_4} =$	Uncontrolled mass emissions of CH4, kg/yr (from equation 4 or LandGEM);
η_{col} =	Efficiency of the landfill gas collector on system, percent; and
2.75 =	Ratio of the molecular weight of CO_2 to the molecular weight of CH_4 .

To prepare estimates of SO_2 emissions, data on the concentration of reduced sulfur compounds within the landfill gas are needed. The best way to prepare this estimate is with site-specific information on the total reduced sulfur content of the landfill gas. Often these data are expressed in ppmv as sulfur (S). Equations 3 and 4 should be used first to determine the uncontrolled mass emission rate of reduced sulfur compounds as sulfur. Then, the following equation can be used to estimate SO_2 emissions:

$$CM_{SO_2} = UM_S \times \frac{\eta_{col}}{100} \times 2.0 \tag{7}$$

where:

CM_{SO_2}	=	Controlled mass emissions of SO ₂ , kg/yr;
UM_S	=	Uncontrolled mass emissions of reduced sulfur compounds as sulfur, kg/y (from
		equations 3 and 4);
η_{col}	=	Efficiency of the landfill gas collection system, percent; and
2.0	=	Ratio of the molecular weight of SO_2 to the molecular weight of S.

The next best method to estimate SO_2 concentrations, if site-specific data for total reduced sulfur compounds as sulfur are not available, is to use site-specific data for speciated reduced sulfur compound concentrations. These data can be converted to ppmv as S with equation 8. After the total reduced sulfur as S has been obtained from equation 8, then equations 3, 4, and 7 can be used to derive SO_2 emissions.

$$C_S = \sum_{i=1}^n C_P \times S_P \tag{8}$$

where:

C_S	=	Concentration of total reduced sulfur compounds, ppmv as S (for use in
		equation 3);
C_P	=	Concentration of each reduced sulfur compound, ppmv;
S_P	=	Number of moles of S produced from the combustion of each reduced sulfur
		compound (e.g., 1 for sulfides, 2 for disulfides); and
n	=	Number of reduced sulfur compounds available for summation.

If no site-specific data are available, a value of 46.9 ppmv can be assumed for C_s (for use in equation 3). This value was obtained by using the default concentrations presented in Table 2.4-1 for reduced sulfur compounds and equation 8.

Hydrochloric acid [Hydrogen Chloride (HCl)] emissions are formed when chlorinated compounds in LFG are combusted in control equipment. The best methods to estimate emissions are mass balance methods that are analogous to those presented above for estimating SO₂ emissions. Hence, the best source of data to estimate HCI emissions is site-specific LFG data on total chloride [expressed in ppmv as the chloride ion (Cl⁻)]. If these data are not available, then total chloride can be estimated from data on individual chlorinated species using equation 9 below. However, emission estimates may be underestimated, since not every chlorinated compound in the LFG will be represented in the laboratory report (i.e., only those that the analytical method specifies).

$$C_{cl} = \sum_{i=1}^{n} C_P \times Cl_P \tag{9}$$

where:

C_{Cl}	=	Concentration of total chloride, ppmv as Cl ⁻ (for use in equation 3);
C_P	=	Concentration of each chlorinated compound, ppmv;
Cl_P	=	Number of moles of Cl ⁻ produced from the combustion of each chlorinated
		compound (e.g., 3 for 1,1,1-trichloroethane); and
n	=	Number of chlorinated compounds available for summation.

After the total chloride concentration (C_{cl}) has been estimated, equations 3 and 4 should be used to determine the total uncontrolled mass emission rate of chlorinated compounds as chloride ion (UM_{Cl}). This value is then used in equation 10 below to derive HCl emission estimates:

$$CM_{HCl} = UM_{Cl} \times \frac{\eta_{col}}{100} \times 1.03 \times \frac{\eta_{cnt}}{100}$$
(10)

where:

CM_{HCl} =	Controlled mass emissions of HCl, kg/yr;
UM_{Cl} =	Uncontrolled mass emissions of chlorinated compounds as chloride, kg/yr (from
	equations 3 and 4);
η_{col} =	Efficiency of the landfill gas collection system, percent;
1.03 =	Ratio of the molecular weight of HCl to the molecular weight of Cl ⁻ ; and
η_{cnt} =	Control efficiency of the landfill gas control or utilization device, percent.

In estimating HCl emissions, it is assumed that all of the chloride ion from the combustion of chlorinated LFG constituents is converted to HCl. If an estimate of the control efficiency, η_{cnt} , is not available, then the high end of the control efficiency range for the equipment listed in Table 2.4-3 should be used. This assumption is recommended to assume that HCl emissions are not under-estimated.

If site-specific data on total chloride or speciated chlorinated compounds are not available, then a default value of 42.0 ppmv can be used for C_{Cl}. This value was derived from the default LFG constituent concentrations presented in Table 2.4-1. As mentioned above, use of this default may produce underestimates of HCl emissions since it is based only on those compounds for which analyses have been performed. The constituents listed in Table 2.4-1 are likely not all of the chlorinated compounds present in LFG.

2.4.5 Source Classification Codes

The Source Classification Codes for Municipal Solid Waste Landfills are:

- 50100401 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Unpaved Road Traffic
- 50100402 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Fugitive Emissions
- 50100403 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Area Method
- 50100404 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Trench Method
- 50100405 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Ramp Method
- 50100406 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Gas Collection System: Other
- 50100407 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Storage Piles
- 50100408 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Conveying of Cover material
- 50100409 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Spreading of Daily Cover
- 50100410 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Dump: Waste Gas Destruction: Waste Gas Flares
- 50100411 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Gas (LFG) Destruction: Incinerator
- 50100412 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Gas (LFG) Destruction: Other Not Elsewhere Classified
- 50100420 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Turbine
- 50100421 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Internal Combustion Engine
- 50100422 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Other Not Elsewhere Classified
- 50100423 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Boiler
- 50100424 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Microturbine
- 50100425 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Direct Use
- 50100426 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Industrial Use
- 50100427 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Automobile Fuel
- 50100433 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Gas (LFG) Purification
- 50100440 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Bioreactor

- 50100441 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Final Cover
- 50100442 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Hazardous Fugitive Emissions
- 50300601 Waste Disposal; Solid Waste Disposal Industrial; Solid Waste Landfill; Waste Gas Destruction
- 50300602 Waste Disposal; Solid Waste Disposal Industrial; Solid Waste Landfill; Other Not Elsewhere Classified
- 50300603 Waste Disposal; Solid Waste Disposal Industrial; Solid Waste Landfill; Hazardous; Fugitive Emissions
- 50300604 Waste Disposal; Solid Waste Disposal Industrial; Solid Waste Landfill; Hazardous Fugitive Emissions
- 50300607 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Storage Piles
- 50300608 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Conveying of Cover material
- 50300609 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Spreading of Daily Cover
- 50301001 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Unpaved Road Traffic
- 50301002 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Area method
- 50301003 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Trench Method
- 50301004 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Ramp Method
- 50301005 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Gas Collection System: Other Not Elsewhere Classified
- 50301010 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Destruction: Incinerator
- 50301011 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Destruction: Other Not Elsewhere Classified
- 50301020 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Turbine
- 50301021 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Internal Combustion Engine
- 50301022 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Other Not Elsewhere Classified
- 50301023 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Boiler
- 50301024 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Microturbine
- 50301025 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Direct Use
- 50301026 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Industrial Use
- 50301027 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Automobile Fuel
- 50301030 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Purification
- 50301040 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Bioreactor
- 50301041 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Final Cover
- 50600601 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Fugitive Emissions
- 50600602 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Hazardous Fugitive Emissions
- 50600603 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Unpaved Road Traffic Area Method
- 50600604 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Area Method
- 50600605 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Trench Method
- 50600606 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Ramp Method
- 50600607 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Gas Collection

System: Other Not Elsewhere Classified

- 50600610 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Destruction: Incinerator
- 50600611 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Destruction: Other Not Elsewhere Classified
- 50600620 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Turbine
- 50600621 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Internal Combustion Engine
- 50600622 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Other Not Elsewhere Classified
- 50600623 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Boiler
- 50600624 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Microturbine
- 50600625 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Direct Use
- 50600626 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Industrial Use
- 50600627 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Automobile Fuel
- 50600630 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Purification
- 50600640 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Bioreactor
- 50600641 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Final Cover
- 50600642 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Conveying of Cover Material
- 50600643 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Storage Piles
- 50600644 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Spreading of Daily Cover
- 50700601 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Fugitive Emissions
- 50700602 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Hazardous Fugitive Emissions
- 50700603 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Unpaved Road Traffic Area Method
- 50700604 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Area Method
- 50700605 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Trench Method
- 50700606 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Ramp Method
- 50700607 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Gas Collection System: Other Not Elsewhere Classified
- 50700610 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Landfill Gas (LFG) Destruction: Incinerator
- 50700611 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Landfill Gas (LFG) Destruction: Other Not Elsewhere Classified
- 50700620 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Turbine
- 50700621 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Internal Combustion Engine
- 50700622 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Other Not Elsewhere Classified
- 50700623 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Boiler

- 50700624 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Microturbine
- 50700625 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Direct Use
- 50700626 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Industrial Use
- 50700627 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Automobile Fuel
- 50700630 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Landfill Gas (LFG) Purification
- 50700640 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Bioreactor
- 50700641 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Final Cover
- 50700642 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Conveying of Cover Material
- 50700643 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Storage Piles
- 50700644 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Spreading of Daily Cover

2.4.6 Major Updates Since the Fifth Edition

August 1998 (Supplement D):

- The equations to calculate the CH₄, CO₂ and other constituents were simplified.
- The default L_0 and k were revised based upon an expanded base of gas generation data.
- The default ratio of CO₂ to CH₄ was revised based upon averages observed in available source test reports.
- The default concentrations of LFG constituents were revised based upon additional data.
- Additional control efficiencies were included, and existing efficiencies were revised based upon additional emission test data.
- The recommended emission factors for secondary compounds emitted from typical control devices were revised and expanded.

November 1998 (Supplement E):

- A correction was made to equation 10.
- Minor changes in the molecular weights for 1,1,1-Trichloroethane (methyl chloroform), 1,1-Dichloroethane, 1,2-Dichloropropane, and Trichloroethylene (trichloroethene) presented in Table 2.4-1 were made to agree with values presented in Perry's Handbook.⁴⁰

August 2024:

- A disclaimer was added on the use of AP-42 emission factors.
- Equation 1 was replaced by Equation HH-1 from 40 CFR 98.343(a)(1) and a conversion factor so that consistent values for methane generation are provided across programs.
- Equation 3 was updated to account for a site-specific fraction of volume of methane in landfill gas.
- The default average collection efficiency was changed from 75% to referring the reader to <u>Table HH-3 to</u> <u>Subpart HH of Part 98 - Landfill Gas Collection Efficiencies</u> for calculations.
- The complete list of Source Classification Codes (SCCs) for MSW landfills were specified and updated in the chapter.
- LandGEM was revised to include Equation HH-1 and its parameter choices.
- SCCs were updated for Tables 2.4-1 2.4-5.
- Default concentrations were rounded to two significant figures in Tables 2.4-1 and 2.4-2 to be consistent with the rest of the chapter and with LandGEM.
- The default concentration for CO was updated (Table 2.4-1).
- NMOC default concentrations were split into pre 1992 and 1992 or later (1992+) to account for codisposal in landfills in Table 2.4-2 and added an additional default concentration was added for NMOC 1992+.

- Affirmed that NMOC is a good surrogate for VOC emissions when NMOC compounds with negligible photochemical reactivity are low, removed default reference to 39% ratio of VOC to NMOC compounds for non-regulatory programs, and reminded users to develop their own ratios or to use extreme caution if default values were selected (footnote c in Table 2.4-2).
- Tables 2.4-4 and 2.4-5 were rearranged and factors for CO from flares and internal combustion engines have been updated, the NO₂ factors from flares and internal combustion engines have been replaced with factors for NO_x from flares and internal combustion engines, a factor for NMOC, as hexane, for flares has been added and 4 factors for NMOC, as hexane, varying by percent load have been added for internal combustion engines.
- Footnote e in Table 2.4-4 was corrected.
- New quality ratings have been given to new/revised factors based on approaches contained in the revised <u>Emissions Factors Procedures Document</u> (January 2023). Factors are given quality ratings based on representativeness of factor (e.g., Highly, Moderately, or Minimally Representative).

1,1,1-Trichloroethane (methyl chloroform)* 133.41 0.48 B 1,1,2,2-Tetrachloroethane* 167.85 1.1 C 1,1-Dichloroethane (ethylidene dichloride)* 98.97 2.4 B 1,1-Dichloroethane (ethylidene dichloride)* 96.94 0.20 B 1,2-Dichloroethane (ethylene dichloride)* 98.96 0.41 B 1,2-Dichloropropane (propylene dichloride)* 112.99 0.18 D 2-Propanol (isopropyl alcohol) 60.11 50 E Acetone 58.08 7.0 B Acrylonitrile* 53.06 6.3 D Butane 163.83 3.1 C Carbon disulfide* 76.13 0.58 C Carbon tetrachloride* 153.84 4.0x10 ³ B Carbon y sulfide* 60.07 0.49 D Chloroethane (ethyl chloride)* 64.52 1.3 B Chloroethane (ethyl chloride)* 64.52 1.3 B Chloroethane (ethyl chloride)* 12.0-91 16 A <th>Compound</th> <th>Molecular Weight</th> <th>Default Concentration (ppmv)</th> <th>Emission Factor Rating</th>	Compound	Molecular Weight	Default Concentration (ppmv)	Emission Factor Rating
1,1-Dichloroethane (ethylidene dichloride)"98.972.4B1,1-Dichloroethane (vinylidene chloride)"96.940.20B1,2-Dichloropothane (propylene dichloride)"112.990.18D2-Propanol (isopropyl alcohol)60.1150EAcetone58.087.0BAcrylonitrile"53.066.3DBromodichloromethane163.833.1CButane58.125.0CCarbon disulfide"76.130.58CCarbon monoxide"112.560.25CCarbon monoxide"112.560.25CChloroethane (ethyl choide)"112.560.25CChlorodifluoromethane64.521.3BChlorodifluoromethane50.491.2BChlorodifluoromethane64.521.3BChlorodifluoromethane50.491.2BChlorodifluoromethane50.491.2BDichlorofluoromethane102.922.6DDichlorodifluoromethane102.922.6DDichlorodifluoromethane30.07890CEthane30.07800CEthane30.0780CEthane30.0780CEthane30.0780CEthane106.164.6BEthylencen*106.164.6BEthylencen*106.164.6BEthylencen*106.164.6B </td <td>1,1,1-Trichloroethane (methyl chloroform)^a</td> <td>-</td> <td></td> <td>-</td>	1,1,1-Trichloroethane (methyl chloroform) ^a	-		-
1.1-Dichloroethene (vinylidene chloride)*96.940.20B1.2-Dichloroethane (ethylene dichloride)*98.960.41B1.2-Dichloropropane (propylene dichloride)*112.990.18D2-Propanol (isopropyl alcohol)60.1150EAcetone58.087.0BAcrylonitrile*53.066.3DBromodichloromethane163.833.1CButane58.125.0CCarbon disulfide*76.130.58CCarbon disulfide*153.844.0x10°3BCarbon tetrachloride*153.844.0x10°3BCarbon tetrachloride*153.844.0x10°3BCarbon tetrachloride*122.560.25CChlorobenzene*112.560.25CChloroforma*119.393.0x10°2BChloroforma*119.393.0x10°2BChlorobenzene*102.9116ADichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.91 </td <td>1,1,2,2-Tetrachloroethane^a</td> <td>167.85</td> <td>1.1</td> <td>С</td>	1,1,2,2-Tetrachloroethane ^a	167.85	1.1	С
1.2-Dichloroethane (ethylene dichloride)*98.960.41B1.2-Dichloropropane (propylene dichloride)*112.990.18D2-Propanol (isopropyl alcohol)60.1150EAcetone58.087.0BAcrylonitrile*53.066.3DBromodichloromethane163.833.1CButane58.125.0CCarbon disulfide*76.130.58CCarbon monoxide*28.01110*Minimally Representative!Carbon tetrachloride*153.844.0x10*BCarbony sulfide*60.070.49DChlorobenzene*112.560.25CChlorobethane86.471.3CChlorobethane50.491.2BDichlorofm*119.393.0x10*2BChlorobethane102.9116ADichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane30.07890CEthane30.07890CEthane30.07890CEthylene dibromide187.881.0x10*3EEthylene dibromide187.881.0x10*3EEthylene dibromide187.881.0x10*3EHuorotrichloromethane106.164.6BEthylene dibromide187.881.0x10*3EHuorotrichlor	1,1-Dichloroethane (ethylidene dichloride) ^a	98.97	2.4	В
1.2-Dichloropropane (propylene dichloride) ^a 112.99 0.18 D 2-Propanol (isopropyl alcohol) 60.11 50 E Acetone 58.08 7.0 B Acetone 53.06 6.3 D Bromodichloromethane 163.83 3.1 C Butane 58.12 5.0 C Carbon disulfide ^a 76.13 0.58 C Carbon monoxide ^b 28.01 110 ^a Minimally Representative ^f Carbon tetrachloride ^a 153.84 4.0x10 ⁻³ B Carbonyl sulfide ^a 60.07 0.49 D Chlorobenzene ^a 112.56 0.25 C Chlorodifluoromethane 86.47 1.3 B Chloroform ^a 119.39 3.0x10 ² B Dichlorobenzene ^c 147 0.21 E Dichlorobenthane 102.92 2.6 D Dichlorobenzene ^c 147 0.21 E Dichlorobenzene ^c 12.91 16 <td< td=""><td>1,1-Dichloroethene (vinylidene chloride)^a</td><td>96.94</td><td>0.20</td><td>В</td></td<>	1,1-Dichloroethene (vinylidene chloride) ^a	96.94	0.20	В
2-Propanol (isopropyl alcohol)60.1150EAcetone58.087.0BActrylonitrilea53.066.3DBromodichloromethane163.833.1CButane58.125.0CCarbon disulfidea76.130.58CCarbon monoxideb28.01110eMinimally RepresentativefCarbon tetrachloridea153.844.0x10 ⁻³ BCarbon tetrachloridea60.070.49DChlorobenzenea112.560.25CChlorodifluoromethane86.471.3CChlorodthane50.491.2BChlorodthane50.491.2BChlorodthane50.491.2BChlorodthane102.922.6DDichlorodfluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102	1,2-Dichloroethane (ethylene dichloride) ^a	98.96	0.41	В
Acetone58.087.0BActrylonitrile³53.066.3DBromodichloromethane163.833.1CButane58.125.0CCarbon disulfideª76.130.58CCarbon monoxide ^b 28.01110°Minimally RepresentativerCarbon tetrachlorideª153.844.0x10³BCarbonyl sulfideª60.070.49DChlorobenzene³112.560.25CChlorodifluoromethane86.471.3CChloroform³119.393.0x10²BChlorodifluoromethane50.491.2BDichlorobenzene ^c 1470.21EDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichloromethane102.933.90CEthanol46.082.7EEthanol45.132.3 </td <td>1,2-Dichloropropane (propylene dichloride)^a</td> <td>112.99</td> <td>0.18</td> <td>D</td>	1,2-Dichloropropane (propylene dichloride) ^a	112.99	0.18	D
Acrylonitrile*53.066.3DBromodichloromethane163.833.1CButane58.125.0CCarbon disulfide*76.130.58CCarbon monoxide*28.01110°Minimally RepresentativefCarbon tetrachloride*153.844.0x10*3BCarbon tetrachloride*60.070.49DChlorobenzene*112.560.25CChlorodifluoromethane86.471.3BChlorodifluoromethane86.471.3BChlorodifluoromethane119.393.0x10*2BChlorodifluoromethane120.911.6ADichlorodifluoromethane120.9116ADichlorodifluoromethane102.922.6DDichlorodifluoromethane30.07890CEthane30.07890CEthane30.07890CEthanel106.164.6BEthyleneare*106.164.6BEthyleneare*106.164.6BEthyleneare*106.164.6BHyleneare*137.380.76BHylorgen sulfide34.0836B	2-Propanol (isopropyl alcohol)	60.11	50	E
Bromodichloromethane163.833.1CButane58.125.0CCarbon disulfide³76.130.58CCarbon monoxide⁵28.01110°Minimally Representative⁵Carbon tetrachloride³153.844.0x10³BCarbon tetrachloride³60.070.49DChlorobenzene³112.560.25CChlorodifluoromethane86.471.3CChlorodifluoromethane64.521.3BChlorodifluoromethane50.491.2BChlorodifluoromethane102.9116ADichlorodifluoromethane102.922.6DDichlorodifluoromethane102.922.6DDichlorodifluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane102.922.6DEthane30.07890CEthane106.164.6BEthylen	Acetone	58.08	7.0	В
Butane58.125.0CCarbon disulfide³76.130.58CCarbon monoxideb28.01110°Minimally RepresentativefCarbon tetrachloride³153.844.0x10 ⁻³ BCarbonyl sulfideª60.070.49DChlorobenzene³112.560.25CChlorodifluoromethane86.471.3CChlorodifluoromethane64.521.3BChlorobenzenec119.393.0x10 ⁻² BChlorodifluoromethane50.491.2BDichlorobenzenec1470.21EDichlorodifluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane30.07890CEthane30.07890CEthane106.164.6BEthyl mercaptan (ethanethiol)62.132.3DEthylene dibromide187.881.0x10 ⁻³ EFluorotrichloromethane137.380.76BHexane ^a 86.186.6BHydrogen sulfide34.0836B	Acrylonitrile ^a	53.06	6.3	D
Carbon disulfide*76.130.58CCarbon monoxideb28.01110°Minimally RepresentativefCarbon tetrachloride*153.844.0x10 ⁻³ BCarbonyl sulfide*60.070.49DChlorobenzene*112.560.25CChlorodifluoromethane86.471.3CChlorodifluoromethane86.471.3BChlorodifluoromethane64.521.3BChlorodifluoromethane50.491.2BChlorodifluoromethane102.9116ADichlorodifluoromethane102.922.6DDichlorofuromethane102.922.6DDichlorofuromethane30.07890CEthane30.07890CEthanel106.164.6BEthylencapta106.164.6BEthylenzene*106.164.6BEthylene dibromide187.881.0x10 ⁻³ EFluorotrichloromethane137.380.76BHexane*86.186.6BHydrogen sulfide34.0836B	Bromodichloromethane	163.83	3.1	С
Carbon monoxideb28.01110°Minimally RepresentativefCarbon tetrachloridea153.844.0x10 ⁻³ BCarbonyl sulfidea60.070.49DChlorobenzenea112.560.25CChlorodifluoromethane86.471.3CChloroethane (ethyl chloride)a64.521.3BChloroethane (ethyl chloride)a50.491.2BChloromethane50.491.2BChlorodifluoromethane120.9116ADichlorobenzenec1470.21EDichlorodifluoromethane102.922.6DDichloroffluoromethane102.922.6DDichloroffluoromethane102.922.6CDichloroffluoromethane30.07890CEthane30.07890CEthane106.164.6BEthyl mercaptan (ethanethiol)62.132.3DEthylenzenea106.164.6BEthylenzenea106.164.6BEthylenzenea137.380.76BHexanea86.186.6BHydrogen sulfide34.0836B	Butane	58.12	5.0	С
Carbon tetrachloride®153.844.0x10°3Representative¹Carbonyl sulfide®60.070.49DChlorobenzene®112.560.25CChlorodifluoromethane86.471.3CChlorodifluoromethane64.521.3BChlorodifluoromethane50.491.2BChlorodifluoromethane50.491.2BDichlorobenzene ^c 1470.21EDichlorodifluoromethane102.922.6DDichlorofluoromethane102.922.6DDichlorofluoromethane30.07890CEthane30.07890CEthanel106.164.6BEthylenzene ^a 106.164.6BEthylene dibromide187.881.0x10°3EFluorotrichloromethane137.380.76BHexane ^a 86.186.6BHydrogen sulfide34.0836B	Carbon disulfide ^a	76.13	0.58	С
Carbon tetrachloridea153.844.0x10-3BCarbonyl sulfidea60.070.49DChlorobenzenea112.560.25CChlorodifluoromethane86.471.3CChlorodifluoromethane64.521.3BChlorodifna119.393.0x10 ⁻² BChlorodenzenec1470.21EDichlorodifluoromethane102.922.6DDichloroffluoromethane102.922.6DDichloroffluoromethane102.922.6DDichloroffluoromethane102.922.6DDichloroffluoromethane102.922.6DDichloroffluoromethane102.922.6DDichloroffluoromethane102.922.6DDichloroffluoromethane102.922.6DDichloroffluoromethane102.922.6DDichloroffluoromethane102.922.6DDichloroffluoromethane102.922.6DDichloroffluoromethane102.922.6DDichloroffluoromethane102.922.6DDichloroffluoromethane102.922.6DDichloroffluoromethane102.922.6DDichloroffluoromethane102.922.6DDichloroffluoromethane102.922.6DEthanol46.0827EEthyl mercaptan (ethanethiol)62.132.3DEthylene dibromide187.881.0x10 ⁻³ E </td <td>Carbon monoxide^b</td> <td>28.01</td> <td>110^e</td> <td>-</td>	Carbon monoxide ^b	28.01	110 ^e	-
Chlorobenzene ^a 112.560.25CChlorodifluoromethane86.471.3CChlorodifluoromethane64.521.3BChloroform ^a 119.393.0x10 ⁻² BChloroform ^a 119.393.0x10 ⁻² BChloromethane50.491.2BDichlorobenzene ^c 1470.21EDichloroffluoromethane102.9116ADichlorofluoromethane102.922.6DDichlorofluoromethane (methylene chloride) ^a 84.9414ADimethyl sulfide (methyl sulfide)62.137.8CEthane30.07890CEEthyl mercaptan (ethanethiol)62.132.3DEthylenea ^a 106.164.6BEthylene dibromide187.881.0x10 ⁻³ EFluorotrichloromethane137.380.76BHexane ^a 86.186.6BHydrogen sulfide34.0836B	Carbon tetrachloride ^a	153.84	4.0x10 ⁻³	
Chlorodifluoromethane86.471.3CChlorodifluoromethane (ethyl chloride)a64.521.3BChloroforma119.393.0x10 ⁻² BChloromethane50.491.2BDichlorobenzene ^c 1470.21EDichloroffluoromethane120.9116ADichloroffluoromethane102.922.6DDichlorofluoromethane (methylene chloride)a84.9414ADimethyl sulfide (methyl sulfide)62.137.8CEthane30.07890CEEthyl mercaptan (ethanethiol)62.132.3DEthyleneaa106.164.6BEFluorotrichloromethane137.380.76BHexanea86.186.6BHydrogen sulfideHydrogen sulfide34.0836B	Carbonyl sulfide ^a	60.07	0.49	D
Chloroethane (ethyl chloride)a64.521.3BChloroforma119.393.0x10 ⁻² BChloromethane50.491.2BDichlorobenzenec1470.21EDichlorodifluoromethane120.9116ADichlorofluoromethane102.922.6DDichlorofluoromethane (methylene chloride)a84.9414ADimethyl sulfide (methyl sulfide)62.137.8CEthane30.07890CEEthyl mercaptan (ethanethiol)62.132.3DEthyllenzena106.164.6BEthylene dibromide187.881.0x10 ⁻³ EFluorotrichloromethane137.380.76BHexanea86.186.6BHydrogen sulfide34.0836B	Chlorobenzene ^a	112.56	0.25	С
Chloroforma119.39 $3.0x10^{-2}$ BChloromethane 50.49 1.2 BDichlorobenzenec147 0.21 EDichlorodifluoromethane 120.91 16 ADichlorofluoromethane 102.92 2.6 DDichlorofluoromethane (methylene chloride)a 84.94 14 ADimethyl sulfide (methyl sulfide) 62.13 7.8 CEthane 30.07 890 CEEthyl mercaptan (ethanethiol) 62.13 2.3 DEthylbenzenea 106.16 4.6 BEthylene dibromide 187.88 $1.0x10^{-3}$ EFluorotrichloromethane 137.38 0.76 BHexanea 86.18 6.6 BHydrogen sulfide 34.08 36 B	Chlorodifluoromethane	86.47	1.3	С
Chloromethane50.491.2BDichlorobenzenec1470.21EDichlorodifluoromethane120.9116ADichlorofluoromethane102.922.6DDichlorofluoromethane (methylene chloride)a84.9414ADimethyl sulfide (methyl sulfide)62.137.8CEthane30.07890CEthanol46.0827EEthyl mercaptan (ethanethiol)62.132.3DEthylbenzenea106.164.6BEthylene dibromide187.881.0x10 ⁻³ EFluorotrichloromethane137.380.76BHexanea86.186.6BHydrogen sulfide34.0836B	Chloroethane (ethyl chloride) ^a	64.52	1.3	В
Dichlorobenzenec1470.21EDichlorodifluoromethane120.9116ADichlorofluoromethane102.922.6DDichloromethane (methylene chloride)a84.9414ADimethyl sulfide (methyl sulfide)62.137.8CEthane30.07890CEthanol46.0827EEthyl mercaptan (ethanethiol)62.132.3DEthylbenzenea106.164.6BEthylene dibromide187.881.0x10 ⁻³ EFluorotrichloromethane137.380.76BHexanea86.186.6BHydrogen sulfide34.0836B	Chloroform ^a	119.39	3.0x10 ⁻²	В
Dichlorodifluoromethane120.9116ADichlorofluoromethane102.922.6DDichloromethane (methylene chloride) ^a 84.9414ADimethyl sulfide (methyl sulfide)62.137.8CEthane30.07890CEthanol46.0827EEthyl mercaptan (ethanethiol)62.132.3DEthylbenzene ^a 106.164.6BEthylene dibromide187.881.0x10 ⁻³ EFluorotrichloromethane137.380.76BHexane ^a 86.186.6BHydrogen sulfide34.0836B	Chloromethane	50.49	1.2	В
Dichlorofluoromethane102.922.6DDichloromethane (methylene chloride)a84.9414ADimethyl sulfide (methyl sulfide)62.137.8CEthane30.07890CEthanol46.0827EEthyl mercaptan (ethanethiol)62.132.3DEthylbenzenea106.164.6BEthylene dibromide187.881.0x10 ⁻³ EFluorotrichloromethane137.380.76BHexanea86.186.6BHydrogen sulfide34.0836B	Dichlorobenzene ^c	147	0.21	E
Dichloromethane (methylene chloride)a84.9414ADimethyl sulfide (methyl sulfide)62.137.8CEthane30.07890CEthanol46.0827EEthyl mercaptan (ethanethiol)62.132.3DEthylbenzenea106.164.6BEthylene dibromide187.881.0x10 ⁻³ EFluorotrichloromethane137.380.76BHexanea86.186.6BHydrogen sulfide34.0836B	Dichlorodifluoromethane	120.91	16	A
Dimethyl sulfide (methyl sulfide)62.137.8CEthane30.07890CEthanol46.0827EEthyl mercaptan (ethanethiol)62.132.3DEthylbenzene ^a 106.164.6BEthylene dibromide187.881.0x10 ⁻³ EFluorotrichloromethane137.380.76BHexane ^a 86.186.6BHydrogen sulfide34.0836B	Dichlorofluoromethane	102.92	2.6	D
Ethane30.07890CEthanol46.0827EEthyl mercaptan (ethanethiol)62.132.3DEthylbenzene ^a 106.164.6BEthylene dibromide187.881.0x10 ⁻³ EFluorotrichloromethane137.380.76BHexane ^a 86.186.6BHydrogen sulfide34.0836B	Dichloromethane (methylene chloride) ^a	84.94	14	A
Ethanol46.0827EEthyl mercaptan (ethanethiol)62.132.3DEthylbenzenea106.164.6BEthylene dibromide187.881.0x10 ⁻³ EFluorotrichloromethane137.380.76BHexanea86.186.6BHydrogen sulfide34.0836B	Dimethyl sulfide (methyl sulfide)	62.13	7.8	С
Ethyl mercaptan (ethanethiol)62.132.3DEthylbenzenea106.164.6BEthylene dibromide187.881.0x10 ⁻³ EFluorotrichloromethane137.380.76BHexanea86.186.6BHydrogen sulfide34.0836B	Ethane	30.07	890	С
Ethylbenzene ^a 106.16 4.6 B Ethylene dibromide 187.88 1.0x10 ⁻³ E Fluorotrichloromethane 137.38 0.76 B Hexane ^a 86.18 6.6 B Hydrogen sulfide 34.08 36 B	Ethanol	46.08	27	E
Ethylene dibromide187.881.0x10 ⁻³ EFluorotrichloromethane137.380.76BHexane ^a 86.186.6BHydrogen sulfide34.0836B	Ethyl mercaptan (ethanethiol)	62.13	2.3	D
Fluorotrichloromethane137.380.76BHexanea86.186.6BHydrogen sulfide34.0836B	Ethylbenzene ^a	106.16	4.6	В
Hexanea86.186.6BHydrogen sulfide34.0836B	Ethylene dibromide	187.88	1.0x10 ⁻³	E
Hydrogen sulfide 34.08 36 B	Fluorotrichloromethane	137.38	0.76	В
	Hexane ^a	86.18	6.6	В
Mercury (total) ^{a,d} 200.61 2.9x10 ⁻⁴ E	Hydrogen sulfide	34.08	36	В
	Mercury (total) ^{a,d}	200.61	2.9x10 ⁻⁴	E

Table 2.4-1. DEFAULT CONCENTRATIONS FOR LFG CONSTITUENTS^a (SCC 50100402, 50300603, 50600601, 50700601)

Table 2.4-1. (Continued)

Compound	Molecular Weight	Default Concentration (ppmv)	Emission Factor Rating
Methyl ethyl ketone ^a	72.11	7.1	А
Methyl isobutyl ketone ^a	100.16	1.9	В
Methyl mercaptan	48.11	2.5	С
Pentane	72.15	3.3	С
Perchloroethylene (tetrachloroethylene) ^a	165.83	3.7	В
Propane	44.09	11	В
t-1,2-dichloroethene	96.94	2.8	В
Trichloroethylene (trichloroethene) ^a	131.4	2.8	В
Vinyl chloride ^a	62.5	7.3	В
Xylenes ^a	106.16	12	В

NOTE: This is not an all-inclusive list of potential LFG constituents, only those for which test data were available at multiple sites. References 41-97. Source Classification Codes in parentheses.

^a Hazardous Air Pollutants listed in Title III of the 1990 Clean Air Act Amendments.

^b Carbon monoxide can exist in LFG, typically in small quantities. This default value should be used with caution. Just 2 of 18 sites showed detectable levels of CO. Note that large values – on the order of 1,000 ppm and greater – can indicate underground combustion or other atypical conditions.

^c Source tests did not indicate whether this compound was the para- or ortho- isomer. The para isomer is a Title III-listed HAP.

^d No data were available to speciate total Hg into the elemental and organic forms.

^e Reference 98.

^f Emission factor is minimally representative of the population. Emission factor quality ratings based on the <u>Emissions Factors</u> <u>Procedures Document</u> (January 2023).

Table 2.4-2. DEFAULT CONCENTRATIONS OF BENZENE, NMOC, AND TOLUENE BASED ON WASTE DISPOSAL HISTORY^a (SCC 50100402, 50300603, 50600601, 50700601)

Pollutant	Molecular	Default	Emission
	Weight	Concentration	Factor Rating
		(ppmv)	
Benzene ^b - Co-disposal	78.11	11	D
Benzene ^b - No or Unknown co-disposal	78.11	1.9	В
NMOC (as hexane) ^c - Co-disposal (pre 1992) ^d	86.18	2400	D
NMOC (as hexane) ^c - No or Unknown co-disposal (pre 1992) ^d	86.18	600	В
NMOC (as hexane) ^c - No or Unknown co-disposal (1992+) ^d	86.18	550°	Moderately Representative ^f
Toluene ^{b-} Co-disposal	92.13	170	D
Toluene ^b - No or Unknown co-disposal	92.13	39	A

^a References 41-85. Source Classification Codes in parentheses.

^b Hazardous Air Pollutants listed in Title III of the 1990 Clean Air Act Amendments.

^c For NSPS/Emission Guideline compliance purposes, the default concentration for NMOC as specified in the final rule must be used. For other purposes, users should develop and use their own-site specific data.

^d Emission factors are split into pre 1992 and 1992 or later (1992+) to account for co-disposal in landfills. The 1992+ factor is based on data from landfills that did not accept co-disposed waste and from landfills that opened in 1992 or later because current landfills do not allow co-disposal. The pre 1992 factors are based on data from landfills that opened before and during 1992, when some landfills allowed co-disposal.

^e Reference 113.

^f Emission factor is moderately representative of the population. Emission factor quality ratings based on the <u>Emissions Factors</u> <u>Procedures Document</u> (January 2023).

Page 23 of 34

Control Device	Constituent ^b	Typical Control	Range of	Emission
		Efficiency (%)	Control	Factor
			Efficiency (%)	Rating
Boiler/Steam Turbine (50100423,	NMOC	98	96-99+	D
50301023, 50600623, 50700623)				
Boiler/Steam Turbine (50100423,	Halogenated Species	99.6	87-99+	D
50301023, 50600623, 50700623)				
Boiler/Steam Turbine (50100423,	Non-Halogenated	99.8	67-99+	D
50301023, 50600623, 50700623)	Species			
Flare ^c (50100410, 50300601)	NMOC	99.2	90-99+	В
Flare ^c (50100410, 50300601)	Halogenated Species	98	91-99+	С
Flare ^c (50100410, 50300601)	Non-Halogenated	99.7	38-99+	С
	Species			
Gas Turbine (50100420,	NMOC	94.4	90-99+	E
50301020, 50600620, 50700620)				
Gas Turbine (50100420,	Halogenated Species	99.7	97-99+	E
50301020, 50600620, 50700620)				
Gas Turbine (50100420,	Non-Halogenated	98.2	97-99+	E
50301020, 50600620, 50700620)	Species			
IC Engine (50100421, 50301021,	NMOC	97.2	94-99+	E
50600621, 50700621)				
IC Engine (50100421, 50301021,	Halogenated Species	93	90-99+	E
50600621, 50700621)				
IC Engine (50100421, 50301021,	Non-Halogenated	86.1	25-99+	E
50600621, 50700621)	Species			

Table 2.4-3. CONTROL EFFICIENCIES FOR LFG CONSTITUENTS^a

^a References 41-97. Source Classification Codes in parentheses.

^b Halogenated species are those containing atoms of chlorine, bromine, fluorine, or iodine. For any equipment, the control efficiency for mercury should be assumed to be 0. See section 2.4.4.2 for methods to estimate emissions of SO₂, CO₂, and HCl.

^c Where information on equipment was given in the reference, test data were taken from enclosed flares. Control efficiencies are assumed to be equally representative of open flares.

Table 2.4-4. (Metric Units) EMISSION FACTORS FOR SECONDARY COMPOUNDS EXITING CONTROL DEVICES^a

Control Device	Pollutant ^b	kg/10 ⁶ dscm Methane	Emission Factor Rating
Boiler/Steam Turbine ^c (50100423, 50301023, 50600623, 50700623)	Nitrogen dioxide	530	D
Boiler/Steam Turbine ^c (50100423, 50301023, 50600623, 50700623)	Carbon monoxide	90	E
Boiler/Steam Turbine ^c (50100423, 50301023, 50600623, 50700623)	Particulate matter	130	D
Gas Turbine (50100420, 50301020, 50600620, 50700620)	Nitrogen dioxide	1,400	D
Gas Turbine (50100420, 50301020, 50600620, 50700620)	Carbon monoxide	3,600	E
Gas Turbine (50100420, 50301020, 50600620, 50700620)	Particulate matter	350	E
Enclosed Combustor/Flared (50100410, 50300601)	Particulate matter	270	D
Enclosed Combustor/Flare ^d (50100410, 50300601)	Nitrogen oxides	610 ^{e,f}	Highly Representative ^g
Enclosed Combustor/Flare ^d (50100410, 50300601)	NMOC, as hexane (VOC) ^j	66 ^{e,h}	Highly Representative ^g
Enclosed Combustor/Flare ^d (50100410, 50300601)	Carbon monoxide	920 ^{e,i}	Highly Representative ^g
IC Engine (50100421, 50301021, 50600621, 50700621)	Nitrogen oxides	1,500 ^{e,k}	Highly Representative ^g
IC Engine (50100421, 50301021, 50600621, 50700621)	Carbon monoxide	4,600 ^{e,I}	Highly Representative ^g
IC Engine (50100421, 50301021, 50600621, 50700621)	Particulate matter	770	E
IC Engine (50100421, 50301021, 50600621, 50700621)	NMOC, as hexane (VOC) ^{j.o} 100% Load	250 ^{e,m}	Moderately Representative ⁿ
IC Engine (50100421, 50301021, 50600621, 50700621)	NMOC, as hexane (VOC) ^{j.o} 80% Load	250 ^{e,m}	Moderately Representative ⁿ
IC Engine (50100421, 50301021, 50600621, 50700621)	NMOC, as hexane (VOC) ^{j.o} 60% Load	270 ^{e,m}	Moderately Representative ⁿ
IC Engine (50100421, 50301021, 50600621, 50700621)	NMOC, as hexane (VOC) ^{j.o} 30% Load	140 ^{e,m}	Moderately Representative ⁿ

^a Source Classification Codes in parentheses. Divide kg/10⁶ dscm methane by 16,666.7 to obtain kg/hr/dscmm methane. ^b No data on PM size distributions were available, however for other gas-fired combustion sources, most of the particulate matter is less than 2.5 microns in diameter. Hence, this emission factor can be used to provide estimates of PM-10 or PM-2.5 emissions. See section 2.4.4.2 for methods to estimate CO₂, SO₂, and HCl. As mentioned in *Basic Information about NO₂*, available at <u>https://www.epa.gov/no2-pollution/basic-information-about-no2</u>, nitrogen dioxide (NO₂) is one of a group of highly reactive gases known as oxides of nitrogen or nitrogen oxides (NO_X), and it is used as the indicator for the larger group of nitrogen oxides. ^c All source tests were conducted on boilers, however emission factors should also be representative of steam turbines. Emission factors are representative of boilers equipped with low-NO_X burners and flue gas recirculation. No data were available for uncontrolled NO_x emissions.

^d Test data were taken from enclosed flares. Control efficiencies are assumed to be equally representative of open flares. ^e Factors were converted from lb/mmbtu. To convert back to lb/mmbtu, divide by 16.02 and then divide by 1020 (the heat content of methane in ft³/Btu). Note that these factors will have units of lb/mmbtu in WebFIRE.

^f Reference 99.

^g Emission factor is highly representative of the population. Emission factor quality ratings based on the <u>Emissions Factors</u> <u>Procedures Document</u> (January 2023).

^h Reference 100.

ⁱ Reference 101.

¹ NMOC = VOC because review of data from references 11-21, 23-24, 26,27, 29,30, 32,36,38,102-109 affirm the effect of compounds with low or no photochemical reactivity is less than 50 ppm LFG.

^k Reference 110.

^I Reference 111.

8/24

^m Reference 112.

ⁿ Emission factor is moderately representative of the population. Emission factor quality ratings based on the <u>Emissions</u> <u>Factors Procedures Document</u> (January 2023).

° During its review, the EPA investigated whether the NMOC emission factors at varying loads could be combined but found that they should remain separate. All per-load heat input values were found to be from differing data sets via a two sample t-test at a 95% confidence coefficient.

Table 2.4-5. (English Units) EMISSION RATES FOR SECONDARY COMPOUNDS EXITING CONTROL DEVICES^a

Control Device	Pollutant ^b	lb/10 ⁶ dscf Methane	Emission Factor Rating
Boiler/Steam Turbine ^c (50100423, 50301023, 50600623, 50700623)	Nitrogen dioxide	33	E
Boiler/Steam Turbine ^c (50100423, 50301023, 50600623, 50700623)	Carbon monoxide	5.7	E
Boiler/Steam Turbine ^c (50100423, 50301023, 50600623, 50700623)	Particulate matter	8.2	E
Gas Turbine (50100420, 50301020, 50600620, 50700620)	Nitrogen dioxide	87	D
Gas Turbine (50100420, 50301020, 50600620, 50700620)	Carbon monoxide	230	D
Gas Turbine (50100420, 50301020, 50600620, 50700620)	Particulate matter	22	E
Enclosed Combustor/Flare ^d (50100410, 50300601)	Particulate matter	17	D
Enclosed Combustor/Flare ^d (50100410, 50300601)	Nitrogen oxides	38 ^{e,f}	Highly Representative ^g
Enclosed Combustor/Flare ^d (50100410, 50300601)	NMOC, as hexane (VOC) ^j	4.1 ^{e,f}	Highly Representative ^g
Enclosed Combustor/Flare ^d (50100410, 50300601)	Carbon monoxide	58 ^{e,f}	Highly Representative ^g
IC Engine (50100421, 50301021, 50600621, 50700621)	Nitrogen oxides	96 ^{e,k}	Highly Representative ^g
IC Engine (50100421, 50301021, 50600621, 50700621)	Carbon monoxide	290 ^{e,I}	Highly Representative ^g
IC Engine (50100421, 50301021, 50600621, 50700621)	Particulate matter	48	E
IC Engine (50100421, 50301021, 50600621, 50700621)	NMOC, as hexane (VOC) ^{j,o} 100% Load	15 ^{e,m}	Moderately Representative ⁿ
IC Engine (50100421, 50301021, 50600621, 50700621)	NMOC, as hexane (VOC) ^{j,o} 80% Load	15 ^{e,m}	Moderately Representative ⁿ
IC Engine (50100421, 50301021, 50600621, 50700621)	NMOC, as hexane (VOC) ^{j,o} 60% Load	17 ^{e,m}	Moderately Representative ⁿ
IC Engine (50100421, 50301021, 50600621, 50700621)	NMOC, as hexane (VOC) ^{j,o} 30% Load	ge,m	Moderately Representative ⁿ

^a Source Classification Codes in parentheses. Divide lb/10⁶ dscf by 16,700 to obtain lb/hr/dscfm.

^b Based on data for other combustion sources, most of the particulate matter will be less than 2.5 microns in diameter. Hence, this emission rate can be used to provide estimates of PM-10 or PM-2.5 emissions. See section 2.4.4.2 for methods to estimate CO₂, SO₂, and HCl. As mentioned in *Basic Information about NO*₂, available at <u>https://www.epa.gov/no2-pollution/basic-information-about-no2</u>, nitrogen dioxide (NO₂) is one of a group of highly reactive gases known as oxides of nitrogen

or nitrogen oxides (NO_X), and it is used as the indicator for the larger group of nitrogen oxides.

^c All source tests were conducted on boilers, however emission factors should also be representative of steam turbines. Emission factors are representative of boilers equipped with low-NO_X burners and flue gas recirculation. No data were available for uncontrolled NO_X emissions.

^d Test data were taken from enclosed flares. Control efficiencies are assumed to be equally representative of open flares. ^e Emission Factors were converted from lb/mmbtu. To convert back to lb/mmbtu, divide by 1020. Note that these factors will have

units of lb/mmbtu in WebFIRE.

^f Reference 99.

^g Emission factor is highly representative of the population. Emission factor quality ratings based on the <u>Emissions Factors</u> <u>Procedures Document</u> (January 2023).

^h Reference 100.

ⁱ Reference 101.

^j NMOC = VOC because review of data from references 11-21, 23-24, 26,27, 29,30, 32,36,38,102-109 affirm the effect of compounds with low or no photochemical reactivity is less than 50 ppm LFG.

^k Reference 110.

Reference 111.

8/24

^m Reference 112.

ⁿ Emission factor is moderately representative of the population. Emission factor quality ratings based on the Emissions Factors Procedures Document (January 2023).

° During its review, the EPA investigated whether the NMOC emission factors at varying loads could be combined but found that they should remain separate. All per-load heat input values were found to be from differing data sets via a two sample t-test at a 95% confidence coefficient.

References for Section 2.4

1. "Criteria for Municipal Solid Waste Landfills," 40 CFR Part 258, Volume 56, No. 196, October 9, 1991.

2. Air Emissions from Municipal Solid Waste Landfills - Background Information for Proposed Standards and Guidelines, Office of Air Quality Planning and Standards, EPA-450/3-90-011a, Chapters 3 and 4, U. S. Environmental Protection Agency, Research Triangle Park, NC, March 1991.

3. Characterization of Municipal Solid Waste in the United States: 1992 Update, Office of Solid Waste, EPA-530-R-92-019, U. S. Environmental Protection Agency, Washington, DC, NTIS No. PB92-207-166, July 1992.

4. Eastern Research Group, Inc., List of Municipal Solid Waste Landfills, Prepared for the U. S. Environmental Protection Agency, Office of Solid Waste, Municipal and Industrial Solid Waste Division, Washington, DC, September 1992.

5. US EPA, Advancing Sustainable Materials Management: 2018 FACT Sheet, December 2020.

6. US EPA, LMOP Landfill and Project Database. <u>https://www.epa.gov/lmop/lmop-landfill-and-project-database</u>.

7. US EPA. Characterization of Municipal Solid Waste in the United States: 1998 Update. US EPA Office of Solid Waste Report No. EPA 530-R-98-007. July 1999.

8. Suggested Control Measures for Landfill Gas Emissions, State of California Air Resources Board, Stationary Source Division, Sacramento, CA, August 1990. (book – Stanford library)- (Reference #4 in chapter 2 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997).*

9. "Standards of Performance for New Stationary Sources and Guidelines for Control of Existing Sources: Municipal Solid Waste Landfills; Proposed Rule, Guideline, and Notice of Public Hearing," 40 CFR Parts 51, 52, and 60, Vol. 56, No. 104, May 30, 1991.

10. S.W. Zison, Landfill Gas Production Curves: Myth Versus Reality, Pacific Energy, City of Commerce, CA, [Unpublished] (Reference #47 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp.4-1, 4-3, 4-6).*

11. TRC Environmental Corporation, TR-145: Compliance Testing of a Landfill Flare at Browning-Ferris Gas Services, Inc.'s Facility in Halifax, Massachusetts, May 1996.

12. Horizon Air Measurement Services, Inc., TR-196: Results of the Biennial Criteria and AB 2588 Air Toxics Source Test on the Simi Valley Landfill Flare, Simi Valley, CA, April 1997.

13. EMCON, TR-258: Source Test Report City of Sacramento Landfill Gas Flare, Sacramento, CA, June, 1996.

14. Blue Sky Environmental, LLC, TR-458: Waste Management: Altamont Landfill Annual Compliance Emissions Test Report # 08097 Landfill Gas Flare - Source A-15, Livermore, CA, September 2008.

15. Blue Sky Environmental, LLC, TR-463: Compliance Source Emissions Test Report # 07115 for Keller Canyon Landfill, Pittsburg, CA, December 2007.

16. Blue Sky Environmental, LLC, TR-464: Source Test Emission Report for One Calledus Flare (A-1) Located at Keller Canyon Landfill, Pittsburg, CA, November 2006.

17. Best Environmental, TR-466: City of Palo Alto Landfill Compliance Emissions Test Report, Palo Alto, CA, October 2008.

18. Best Environmental, TR-467: City of Palo Alto Landfill Compliance Emissions Test Report, Palo Alto, CA, October 2007.

19. Blue Sky Environmental, LLC, TR-468: Compliance Source Emissions Test Report for Republic Services – Potrero Hills Landfill, Inc, Suisun, CA, September 2007.

20. Blue Sky Environmental, LLC, TR-469: Annual Compliance Emissions Test Report #08058 Source Test for Landfill Gas

Flare-Source A-51 for Redwood Landfill, Inc, Novato, CA, July 2008.

21. Best Environmental, TR-470: Vasco Road Landfill Compliance Emissions Test Report, Livermore, CA, October 2006.

22. Horizon Air Measurement Services, Inc., TR-530: Emission Compliance Test on Two Landfill Gas Flares San Marcos Landfill, San Marcos, CA, November 1994.

23. Horizon Air Measurement Services, Inc., TR-534: Emission Compliance Test on a Landfill Gas Flare for Sunshine Canyon Landfill, Sylmar, California, October 2007.

24. Horizon Air Measurement Services, Inc., TR-535: Emission Compliance Test on a Landfill Gas Flare After Louver Altercation for Sunshine Canyon Landfill, Sylmar, California, October 2007.

25. Horizon Air Measurement Services, Inc., TR-538: Emission Compliance Test on a Landfill Gas Flare for Simi Valley Landfill and Recycling Center, Simi Valley, CA, February 2008.

26. Horizon Air Measurement Services, Inc., TR-540: Results of the Criteria and AB 2588 Air Toxics Source Test on Simi Valley Landfill Flare #1 (McGill), Simi Valley, CA, December 2005.

27. Horizon Air Measurement Services, Inc., TR-541: Results of the Criteria and AB 2588 Air Toxics Source Test on Simi Valley Landfill Flare #2 (John Zink), Simi Valley, CA, December 2005.

28. Horizon Air Measurement Services, Inc., TR-546: Emissions Compliance Test Results on a Landfill Gas Flare, Palmdale, CA, March 2006.

29. Horizon Air Measurement Services, Inc., TR-579: Source Evaluation Report Waste Management Disposal Services of Washington, Inc., Greater Wenatchee Regional Landfill and Recycling Center, East Wenatchee, WA, February 2007.

30. Alaska Source Testing, LLC, TR-582: Summary of Test Results Municipality of Anchorage, Solid Waste Services Anchorage Landfill Gas Collection and Control System, Anchorage, AK, January 2007.

31. SCEC, TR-603: Waimanalo Gulch Solid Waste Landfill Flare Compliance Source Test Report 2005, Kapolei, HI, January 2006.

32. Shaw EMCON/OWT, Inc., TR-635: 2005 Annual Source Test Report - Redwood Landfill, Novato, CA, August 2005.

33. Shaw EMCON/OWT, Inc., TR-663: 2005 Annual Source Test Report – Kirby Canyon Recycling and Disposal Facility Landfill Gas Control - Flare, Morgan Hill, CA, February 2005.

34. Blue Sky Environmental, LLC, TR-678: Kirby Canyon Recycling and Disposal Facility Annual Compliance Emissions Test Report #07116 Landfill Gas Flare Source A-11, San Jose, CA, January 2008.

35. Blue Sky Environmental, LLC, TR-679: Kirby Canyon Recycling and Disposal Facility Annual Compliance Emissions Test Report #08004 Initial Source Test for Landfill Gas Flare Source A-12, San Jose, CA, March 2008.

36. Shaw Environmental, Inc., TR-711: 2006 Source Test Report Emissions Monitoring of Two Landfill Gas Fired Flares at the Redwood Landfill, Novato, CA, July 2006.

37. Shaw Environmental, Inc., TR-716: 2007 Source Test Report Emissions Monitoring of Two Landfill Gas Fired Flares at the Redwood Landfill, Novato, CA, May 2007.

38. Shaw Environmental, Inc., TR-718: 2005 Annual Source Test Report for Kirby Canyon Recyling and Disposal Facility, Morgan Hill, CA, February 2006.

39. Letter and attached documents from C. Nesbitt, Los Angeles County Sanitation Districts, to K. Brust, E.H. Pechan and Associates, Inc., December 6, 1996.

40. Perry, J.H., Chemical Engineers' Handbook, New York: McGraw-Hill, 1984.

41. Engineering-Science, Inc., Report of Stack Testing at County Sanitation District Los Angeles Puente Hills Landfill, LosAngeles County Sanitation District, August 15, 1984. (Reference #4 in chapter 4 of Emission Factor Documentation for8/24Solid Waste Disposal2.4-23

AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-5, 4-16, 4-17, 4-18, 4-32).*

42. J.R. Manker, Vinyl Chloride (and Other Organic Compounds) Content of Landfill Gas Vented to an Inoperative Flare, Source Test Report 84-496, David Price Company, South Coast Air Quality Management District, November 30, 1984. (Reference #5 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-5, 4-16, 4-17, 4-18, 4-32, B-1, B-3, B-5, B-6, B-9, B-14, B-16, B-17, B-18, B-20). *

43. S. Marinoff, Landfill Gas Composition, Source Test Report 85-102, Bradley Pit Landfill, South Coast Air Quality Management District, May 22, 1985. (Reference #6 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-5, 4-16, 4-17, 4-18, 4-32, B-1, B-3, B-4, B-6, B-8, B-11, B-12, B-14, B-15, B-17, B-18, B-19, B-21). *

44. J. Littman, Vinyl Chloride and Other Selected Compounds Present in A Landfill Gas Collection System Prior to and after Flaring, Source Test Report 85-369, Los Angeles County Sanitation District, South Coast Air Quality Management District, October 9, 1985. (Reference #7 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-5, 4-16, 4-17, 4-18, 4-32, B-1, B-3, B-4, B-6, B-7, B-8, B-11, B-12, B-14, B-15, B-17, B-18, B-19, B-22). *

45. W.A. Nakagawa, Emissions from a Landfill Exhausting Through a Flare System, Source Test Report 85-461, Operating Industries, South Coast Air Quality Management District, October 14, 1985. (Reference #8 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-5, 4-16, 4-17, 4-18, 4-32, B-5, B-16, B-18, B-20, B-22). *

46. S. Marinoff, Emissions from a Landfill Gas Collection System, Source Test Report 85-511. Sheldon Street Landfill, South Coast Air Quality Management District, December 9, 1985. (Reference #9 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-5, 4-16, 4-17, 4-18, 4-32, B-2, B-5, B-7, B-9, B-14, B-16, B-19, B-20, B-22). *

47. W.A. Nakagawa, Vinyl Chloride and Other Selected Compounds Present in a Landfill Gas Collection System Prior to and after Flaring, Source Test Report 85-592, Mission Canyon Landfill, Los Angeles County Sanitation District, South Coast Air Quality Management District, January 16, 1986. (Reference #10 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-5, 4-16, 4-17, 4-18, 4-32, B-1, B-5, B-6, B-9, B-14, B-18, B-20). *

48.California Air Resources Board, Evaluation Test on a Landfill Gas-Fired Flare at the BKK Landfill Facility, West Covina, CA, ARB-SS-87-09, July 1986. (Reference #12 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-5, 4-16, 4-17, 4-18, 4-33, B-1, B-3, B-4, B-6, B-7, B-8, B-10, B-11, B-12, B-13, B-14, B-15, B-18, B-19, B-22). *

49. S. Marinoff, Gaseous Composition from a Landfill Gas Collection System and Flare, Source Test Report 86-0342, Syufy Enterprises, South Coast Air Quality Management District, August 21, 1986. (Reference #13 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-5, 4-16, 4-17, 4-18, 4-33, B-1, B-4, B-6, B-8, B-14, B-15, B-18, B-19, B-21). *

50. Analytical Laboratory Report for Source Test, Azusa Land Reclamation, June 30, 1983, South Coast Air Quality Management District. (Reference #15 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-5, 4-16, 4-17, 4-18, 4-33, B-1, B-3, B-4, B-6, B-7, B-8, B-10, B-12, B-13, B-15, B-17, B-18, B-19, B-21, C3-Flares-1). *

51. J.R. Manker, Source Test Report C-84-202, Bradley Pit Landfill, South Coast Air Quality Management District, May 25, 1984. (Reference #17 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-5, 4-16, 4-17, 4-18, 4-33, B-1, B-3, B-4, B-14, B-15, B-18, B-19, B-21, B-22, C3-Flares-1). *

52. S. Marinoff, Source Test Report 84-315, Puente Hills Landfill, South Coast Air Quality Management District, February 6, 1985. (Reference #18 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1 4-5, 4-16, 4-17, 4-18, 4-33, B-2, B-3, B-5, B-6, B-9, B-14, B-16, B-17, B-19, B-20, B-21).* 8/24

53. P.P. Chavez, Source Test Report 84-596, Bradley Pit Landfill, South Coast Air Quality Management District, March 11, 1985. (Reference #19 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-5, 4-16, 4-17, 4-18, 4-33, B-1, B-3, B-6, B-8, B-14, B-15, B-17, B-19, B-21). *

54. S. Marinoff, Source Test Report 84-373, Los Angeles By-Products, South Coast air Quality Management District, March 27, 1985. (Reference #20 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-5, 4-16, 4-17, 4-18, 4-33, B-2, B-3, B-5, B-7, B-9, B-14, B-16, B-17, B-19, B-20, B-22). *

55. J. Littman, Source Test Report 85-403, Palos Verdes Landfill, South Coast Air Quality Management District, September 25, 1985. (Reference #22 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-5, 4-16, 4-17, 4-18, 4-33, B-1, B-2, B-3, B-5, B-7, B-9, B-14, B-16, B-18, B-19, B-20, B-22, C3-Flares-1). *

56. S. Marinoff, Source Test Report 86-0234, Pacific Lighting Energy Systems, South Coast Air Quality Management District, July 16, 1986. (Reference #23 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-5, 4-16, 4-17, 4-18, 4-33, B-3, B-6, B-9, B-15, B-16, B-19, B-21). *

57. South Coast Air Quality Management District, Evaluation Test on a Landfill Gas-Fired Flare at the Los Angeles County Sanitation District's Puente Hills Landfill Facility, [ARB/SS-87-06], Sacramento, CA, July 1986. (Reference #24 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-5, 4-16, 4-17, 4-18, 4-33, B-2, B-5, B-6, B-7, B-9, B-11, B-14, B-16, B-19, B-20, B-21) *

58. Browning-Ferris Industries, Source Test Report, Lyon Development Landfill, August 21, 1990. (Reference #27 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-5, 4-16, 4-17, 4-18, 4-34, B-2, B-3, B-5, B-7, B-8, B-9, B-11, B-17, B-18, B-20, B-22). *

59. X.V. Via, Source Test Report, Browning-Ferris Industries, Azusa Landfill. (Reference #28 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-3 data excluded because no data support, 4-34). *

60. M. Nourot, Gaseous Composition from a Landfill Gas Collection System and Flare Outlet. Laidlaw Gas Recovery Systems, to J.R. Farmer, OAQPS: ESD, December 8, 1987. (Reference #41 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-5, 4-16, 4-17, 4-18, 4-35, B-1, B-4, B-6, B-8, B-10, B-11, B-12, B-13, B-14, B-15, B-16, B-17, B-18, B-20, B-22, B-23). *

61. D.A. Stringham and W.H. Wolfe, Waste Management of North America, Inc., to J.R. Farmer, OAQPS: ESD, January 29, 1988, Response to Section 114 questionnaire.

62. V. Espinosa, Source Test Report 87-0318, Los Angeles County Sanitation District Calabasas Landfill, South Coast Air Quality Management District, December 16, 1987. (Reference #48 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-5, 4-16, 4-17, 4-18, 4-35, A-7, C3-Turbines-1). *

63. C.S. Bhatt, Source Test Report 87-0329, Los Angeles County Sanitation District, Scholl Canyon Landfill, South Coast Air Quality Management District, December 4, 1987. (Reference #49 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-5, 4-16, 4-17, 4-18, 4-35, A-8). *

64. V. Espinosa, Source Test Report 87-0391, Puente Hills Landfill, South Coast Air Quality Management District, February 5, 1988. (Reference #50 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-5, 4-16, 4-17, 4-18, 4-35, A-8, B-2, B-5, B-6, B-11, B-13, B-14, B-16, B-19, B-20, B-21, B-23, C3-Engines-1). *

65. V. Espinosa, Source Test Report 87-0376, Palos Verdes Landfill, South Coast Air Quality Management District, February 9, 1987. (Reference #51 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-5, 4-16, 4-17, 4-18, 4-35, A-8, B-2, B-5, B-7, B-13, B-14, B-16, B-19, B-20, B-22, B-23, C3-Engines-1). * 66. Bay Area Air Quality Management District, Landfill Gas Characterization, Oakland, CA, 1988.

67. Steiner Environmental, Inc., Emission Testing at BFI's Arbor Hills Landfill, Northville, Michigan, September 22 through 25, 1992, Bakersfield, CA, December 1992.

68. PEI Associates, Inc., Emission Test Report - Performance Evaluation Landfill-Gas Enclosed Flare, Browning Ferris Industries, Chicopee, MA, 1990.

69. Kleinfelder Inc., Source Test Report Boiler and Flare Systems, Prepared for Laidlaw Gas Recovery Systems, Coyote Canyon Landfill, Diamond Bar, CA, 1991.

70. Bay Area Air Quality Management District, McGill Flare Destruction Efficiency Test Report for Landfill Gas at the Durham Road Landfill, Oakland, CA, 1988.

71. San Diego Air Pollution Control District, Solid Waste Assessment for Otay Valley/Annex Landfill. San Diego, CA, December 1988.

72. PEI Associates, Inc., Emission Test Report - Performance Evaluation Landfill Gas Enclosed Flare, Rockingham, VT, September 1990.

73. Browning-Ferris Industries, Gas Flare Emissions Source Test for Sunshine Canyon Landfill. Sylmar, CA, 1991.

74. Scott Environmental Technology, Methane and Nonmethane Organic Destruction Efficiency Tests of an Enclosed Landfill Gas Flare, April 1992.

75. BCM Engineers, Planners, Scientists and Laboratory Services, Air Pollution Emission Evaluation Report for Ground Flare at Browning Ferris Industries Greentree Landfill, Kersey, Pennsylvania. Pittsburgh, PA, May 1992.

76. EnvironMETeo Services Inc., Stack Emissions Test Report for Ameron Kapaa Quarry, Waipahu, HI, January 1994.

77. Waukesha Pearce Industries, Inc., Report of Emission Levels and Fuel Economies for Eight Waukesha 12V-AT25GL Units Located at the Johnston, Rhode Island Central Landfill, Houston TX, July 19, 1991.

78. Mostardi-Platt Associates, Inc., Gaseous Emission Study Performed for Waste Management of North America, Inc., CID Environmental Complex Gas Recovery Facility, August 8, 1989. Chicago, IL, August 1989.

79. Mostardi-Platt Associates, Inc., Gaseous Emission Study Performed for Waste Management of North America, Inc., at the CID Environmental Complex Gas Recovery Facility, July 12-14, 1989. Chicago, IL, July 1989.

80. Browning-Ferris Gas Services, Inc., Final Report for Emissions Compliance Testing of One Waukesha Engine Generator, Chicopee, MA, February 1994.

81. Browning-Ferris Gas Services, Inc., Final Report for Emissions Compliance Testing of Three Waukesha Engine Generators, Richmond, VA, February 1994.

82. South Coast Environmental Company (SCEC), Emission Factors for Landfill Gas Flares at the Arizona Street Landfill, Prepared for the San Diego Air Pollution Control District, San Diego, CA, November 1992.

83. Carnot, Emission Tests on the Puente Hills Energy from Landfill Gas (PERG) Facility - Unit 400, September 1993, Prepared for County Sanitation Districts of Los Angeles County, Tustin, CA, November 1993.

84. Pape & Steiner Environmental Services, Compliance Testing for Spadra Landfill Gas-to-Energy Plant, July 25 and 26, 1990, Bakersfield, CA, November 1990.

85. AB2588 Source Test Report for Oxnard Landfill, July 23-27, 1990, by Petro Chem Environmental Services, Inc., for Pacific Energy Systems, Commerce, CA, October 1990.

86. AB2588 Source Test Report for Oxnard Landfill, October 16, 1990, by Petro Chem Environmental Services, Inc., for Pacific Energy Systems, Commerce, CA, November 1990.

87. Engineering Source Test Report for Oxnard Landfill, December 20, 1990, by Petro Chem Environmental Services,8/24Solid Waste Disposal2.4-26

Inc., for Pacific Energy Systems, Commerce, CA, January 1991.

88. AB2588 Emissions Inventory Report for the Salinas Crazy Horse Canyon Landfill, Pacific Energy, Commerce, CA, October 1990.

89. Newby Island Plant 2 Site IC Engine's Emission Test, February 7-8, 1990, Laidlaw Gas Recovery Systems, Newark, CA, February 1990.

90. Landfill Methane Recovery Part II: Gas Characterization, Final Report, Gas Research Institute, December 1982.

91. Letter from J.D. Thornton, Minnesota Pollution Control Agency, to R. Myers, U.S. EPA, February 1, 1996.

92. Letter and attached documents from M. Sauers, GSF Energy, to S. Thorneloe, U.S. EPA, May 29, 1996.

93. Landfill Gas Particulate and Metals Concentration and Flow Rate, Mountaingate Landfill Gas Recovery Plant, Horizon Air Measurement Services, prepared for GSF Energy, Inc., May 1992.

94. Landfill Gas Engine Exhaust Emissions Test Report in Support of Modification to Existing IC Engine Permit at Bakersfield Landfill Unit #1, Pacific Energy Services, December 4, 1990. (Reference #98 in chapter 4 of Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997, pp. 4-1, 4-2, 4-5, 4-16, 4-17, 4-18, 4-39, A-21).*

95. Addendum to Source Test Report for Superior Engine #1 at Otay Landfill, Pacific Energy Services, April 2, 1991.

96. Source Test Report 88-0075 of Emissions from an Internal Combustion Engine Fueled by Landfill Gas, Penrose Landfill, Pacific Energy Lighting Systems, South Coast Air Quality Management District, February 24, 1988.

97. Source Test Report 88-0096 of Emissions from an Internal Combustion Engine Fueled by Landfill Gas, Toyon Canyon Landfill, Pacific Energy Lighting Systems, March 8, 1988.

98. Carbon Monoxide, stack test data submitted to CEDRI for SCC/control device combination. Each individual test report can be obtained via WebFIRE.

99. NOx stack test data submitted to CEDRI for SCC/control device combination. Each individual test report can be obtained via WebFIRE.

100. NMOC, as hexane, stack test data submitted to CEDRI for SCC/control device combination. Each individual test report can be obtained via WebFIRE.

101. CO stack test data submitted to CEDRI for SCC/control device combination. Each individual test report can be obtained via WebFIRE.

102. SCEC, TR-461: Allied Waste Forward Inc. Landfill Flare 2006 Source Test Results, Diamond Bar, CA, August 2006.

103. SCEC, TR-462: Compliance Source Test Report Austin Road Landfill, Stockton, CA, June 2007.

104. SCEC, TR-526: Final Source Test Report-Otay Landfill John Zink Flare SCEC Job No. 2170.1005, Chula Vista, CA, September 2004.

105. Horizon Air Measurement Services, TR-543: Results of the Biennial Criteria Source Test on the Simi Valley Landfill Flare No. 1 (McGill), Simi Valley, October 2003.

106. Horizon Air Measurement Services, TR-544: Results of the Biennial Criteria Source Test on the Simi Valley Landfill Flare No. 2 (John Zink), Simi Valley, October 2003.

107. Total Air Analysis, Inc., TR-552: Compliance Source Test Result – Bradley Landfill, Sun Valley, CA, August 2008.

108. Total Air Analysis, Inc., TR-553: Compliance Source Test Result – Bradley Landfill, Sun Valley, CA, July 2008.

109. Blue Sky Environmental, LLC, TR-632: TriCities Recycling Disposal Facility Annual Compliance Emissions Test Report #08071 Source Test for Landfill Gas Flare – Source A-3, Fremont, CA, July 2008.

110. NOx stack test data submitted to CEDRI for SCC/control device combination. Each individual test report can be obtained via WebFIRE.

111. CO stack test data submitted to CEDRI for SCC/control device combination. Each individual test report can be obtained via WebFIRE.

112. NMOC, as hexane, stack test data submitted to CEDRI for SCC/control device combination. Each individual test report can be obtained via WebFIRE.

113. NMOC, as hexane, Flare Inlet stack test data submitted to CEDRI for SCC/control device combination. Each individual test report can be obtained via WebFIRE.

* Test data from these reports are in the Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997 (reference 115). The documentation reference number and page citations in the document are listed. The appendices have data from the reports.

References are available electronically at: https://gaftp.epa.gov/ap42/ch02/s04/reference/2024%20Chapter/.